

# **CHEMISTRY**

---

## **A EUROPEAN JOURNAL**

---

### **Supporting Information**

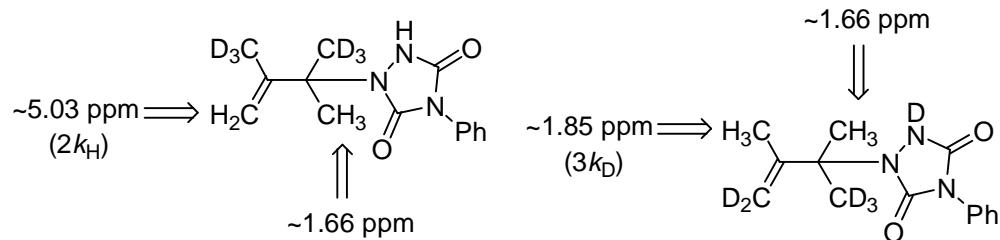
© Copyright Wiley-VCH Verlag GmbH & Co. KGaA, 69451 Weinheim, 2008

# **Solvent-Dependent Changes in the Triazolinedione-Alkene Ene Reaction Mechanism**

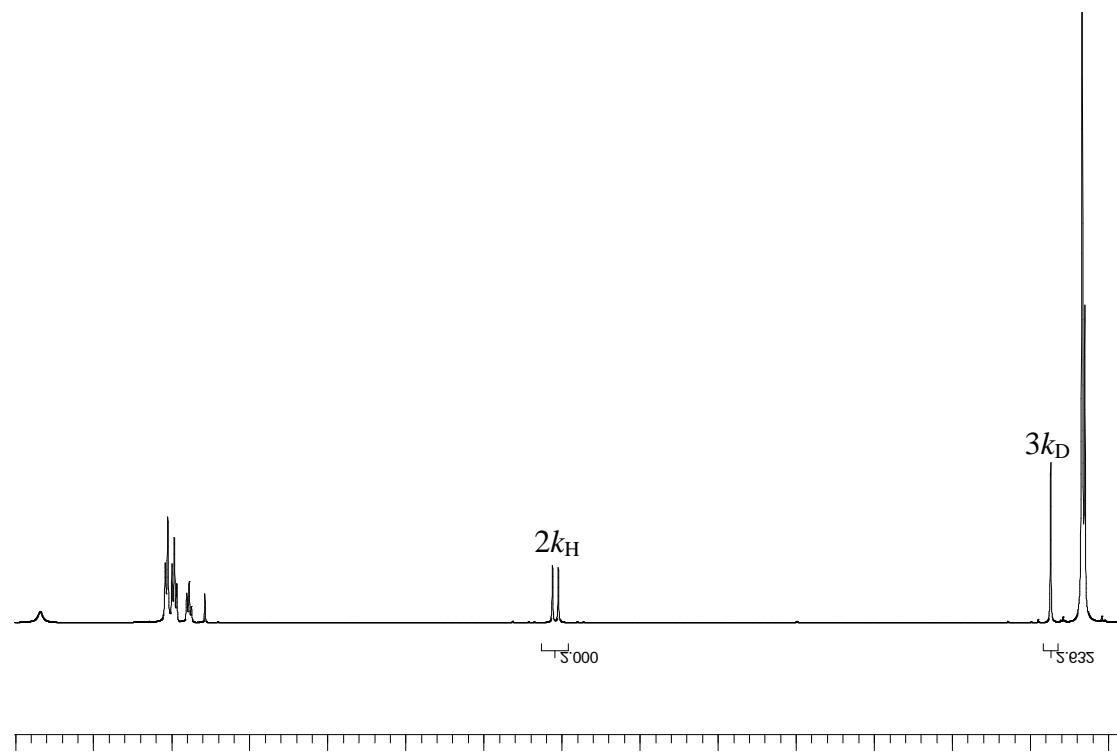
Georgios C. Vougioukalakis, Manolis M. Roubelakis, Mariza N. Alberti, and Michael  
Orfanopoulos\*

*Department of Chemistry, University of Crete, Heraklion, Voutes 71003, Crete, Greece*

**A)  $^1\text{H}$  NMR spectra for the measurement of the KIEs in the reaction of *cis*-TME-*d*6 with PTAD\***

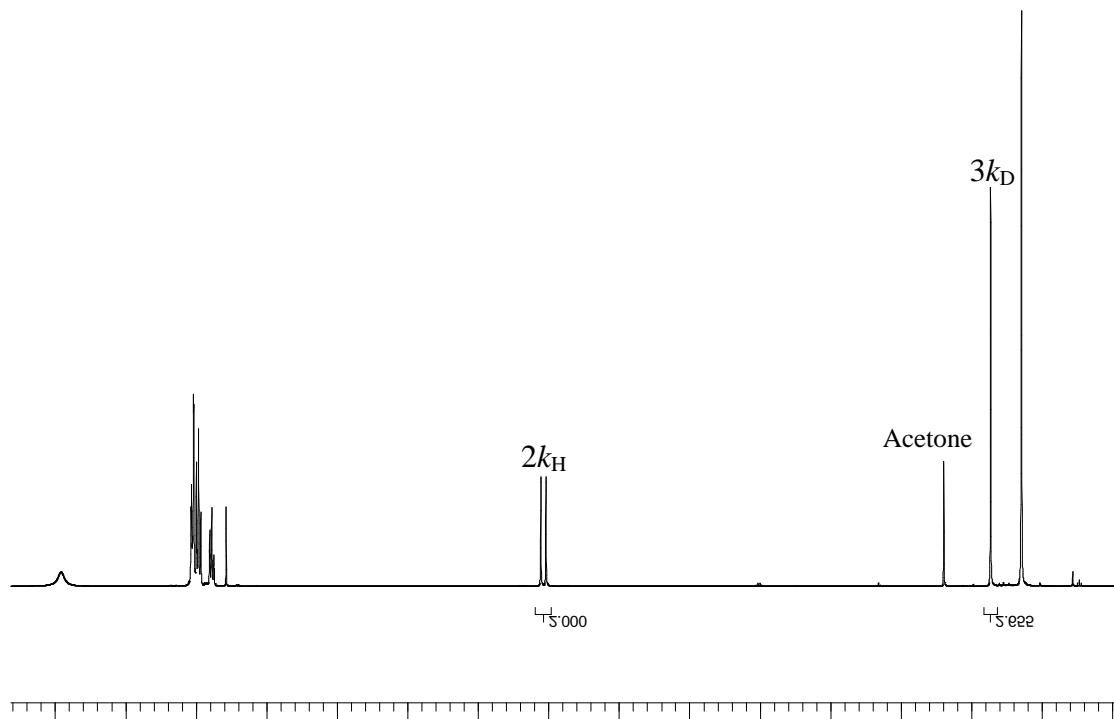


substrate <i>cis</i> -TME- <i>d</i> 6	solvent $\text{CDCl}_3$	temperature rt	$k_{\text{H}}/k_{\text{D}}$ $1.14 \pm 0.03$
--	----------------------------	-------------------	--

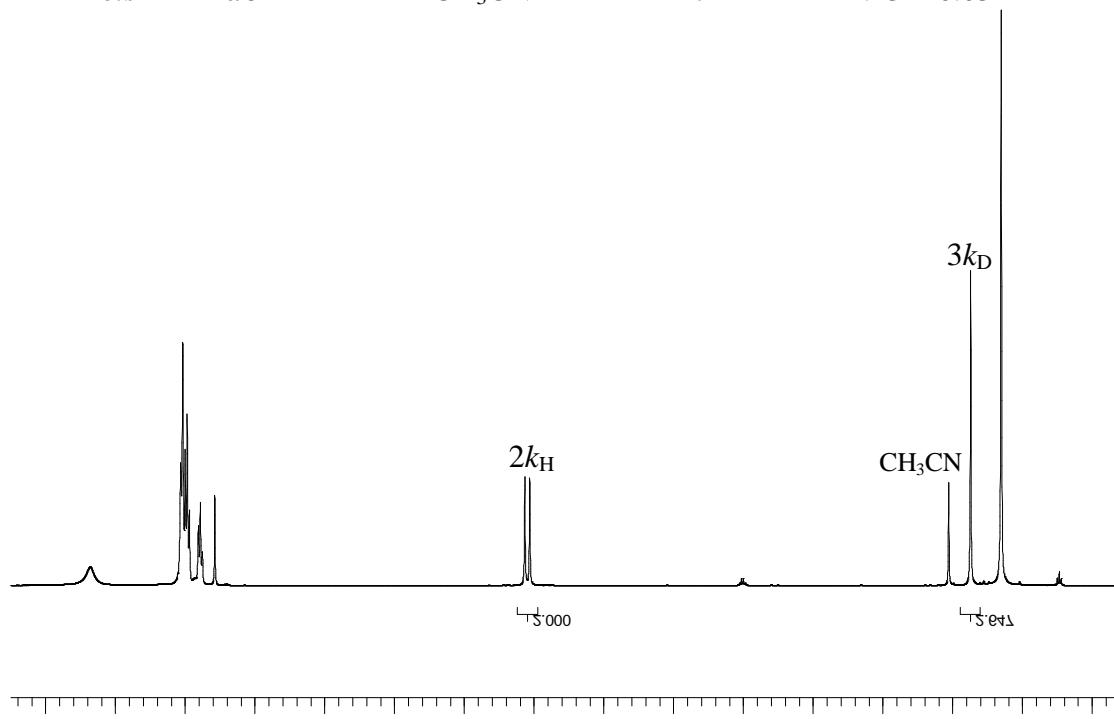


\* In all  $^1\text{H}$  NMR integrations a spin-lattice relaxation time  $T_1 = 5\text{sec}$  was used.

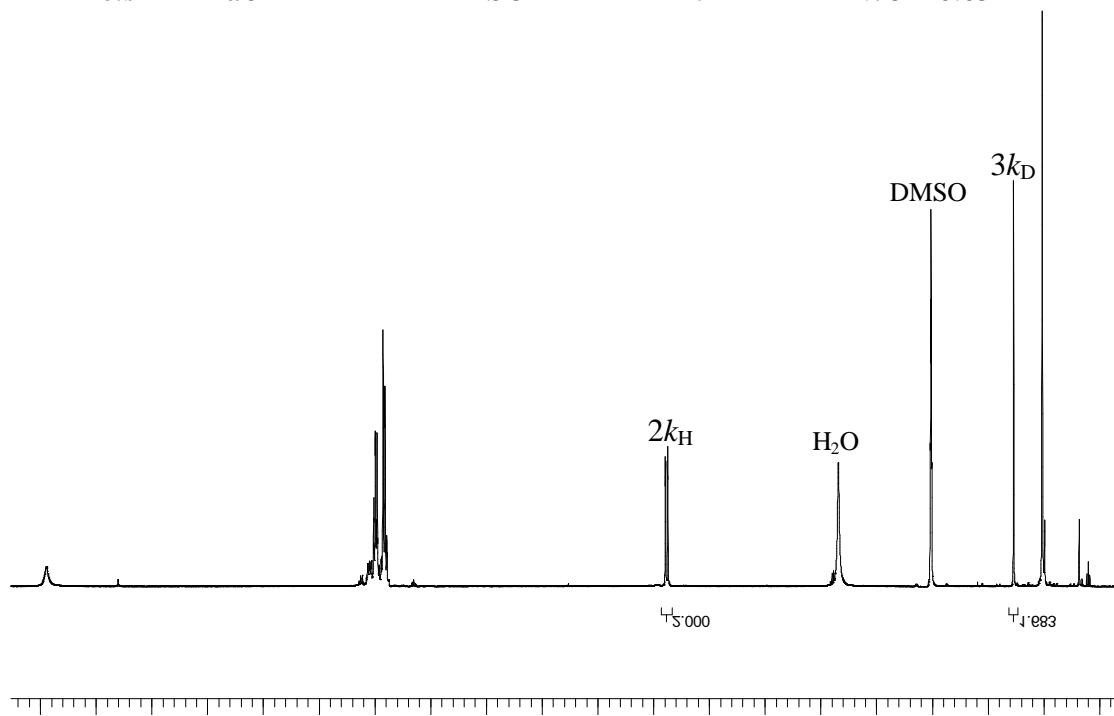
substrate solvent temperature  $k_H/k_D$   
*cis*-TME-*d6* acetone rt  $1.13 \pm 0.03$



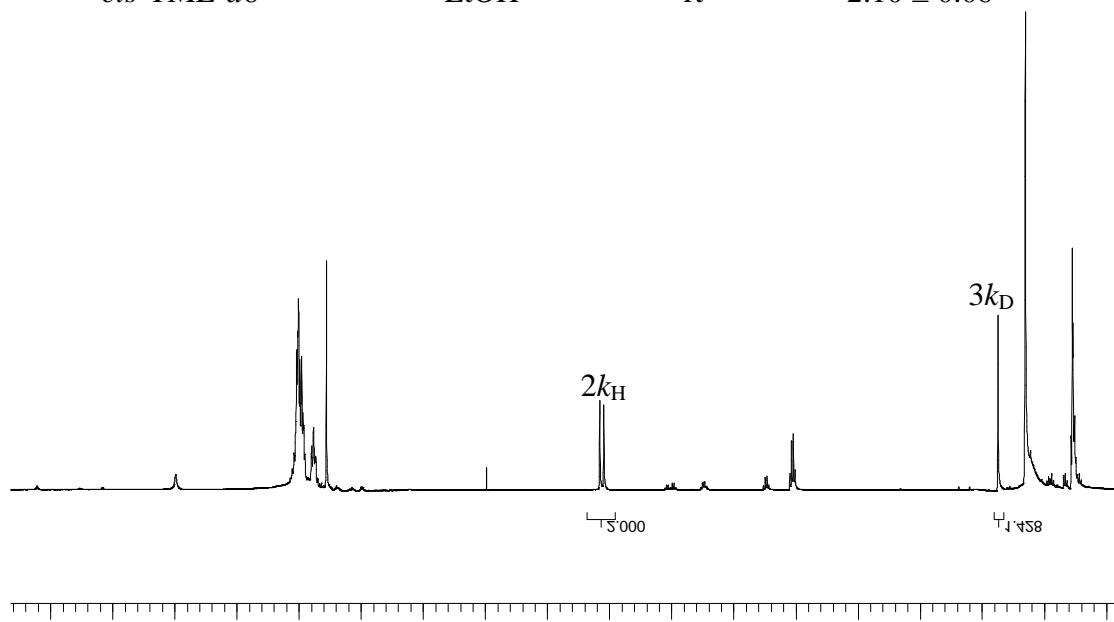
substrate solvent temperature  $k_H/k_D$   
*cis*-TME-*d6* CH<sub>3</sub>CN rt 1.13 ± 0.03

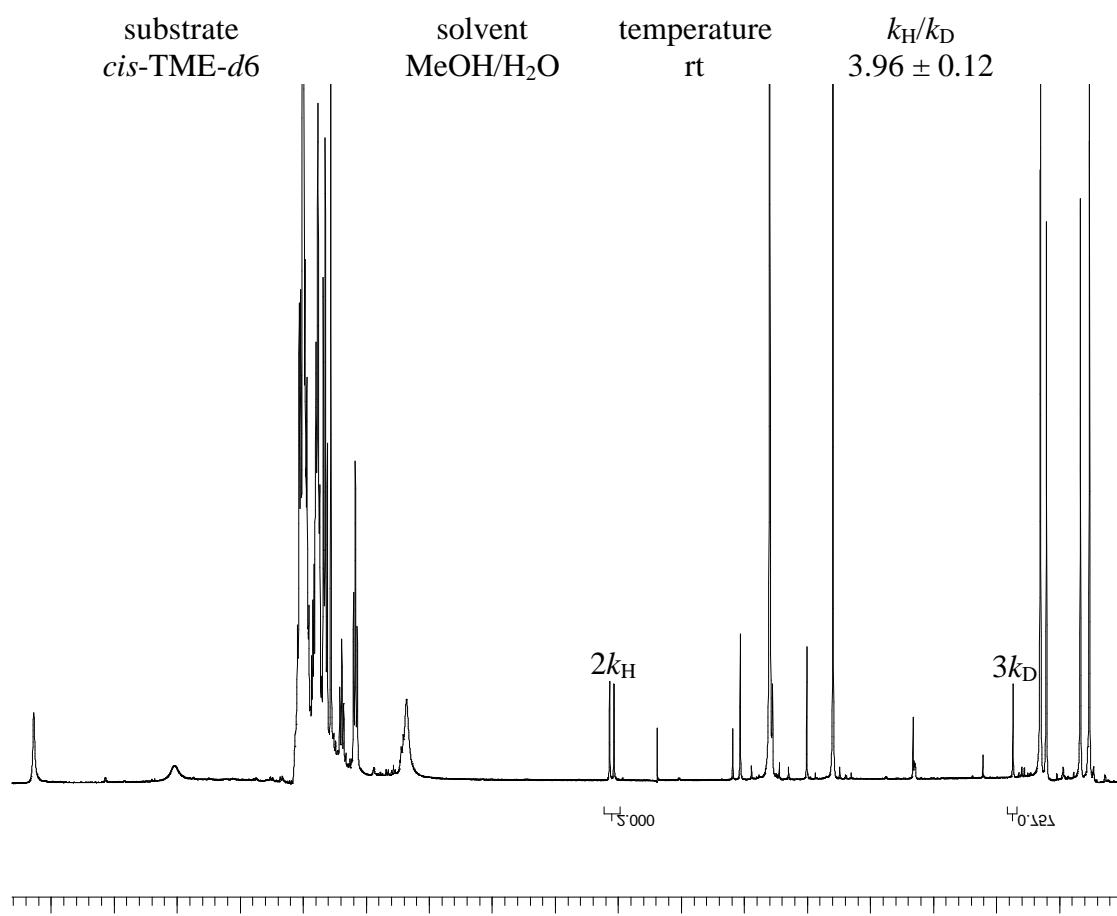
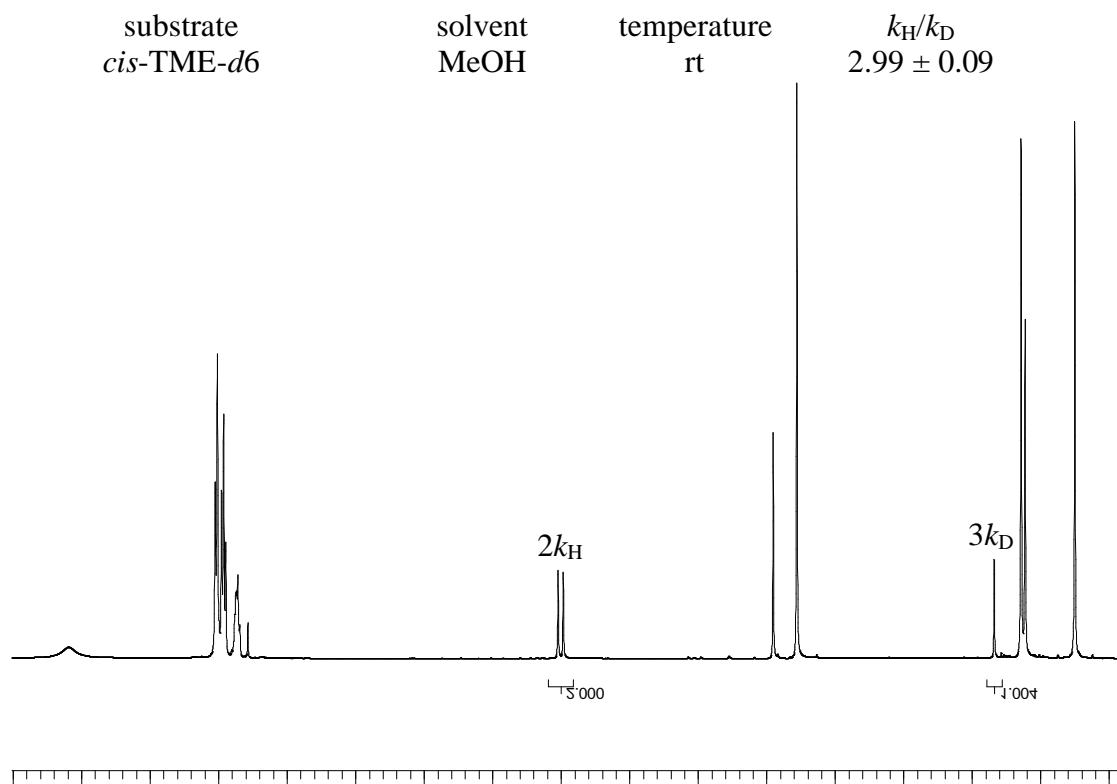


substrate solvent temperature  $k_H/k_D$   
*cis*-TME-*d*6 DMSO rt  $1.78 \pm 0.05$

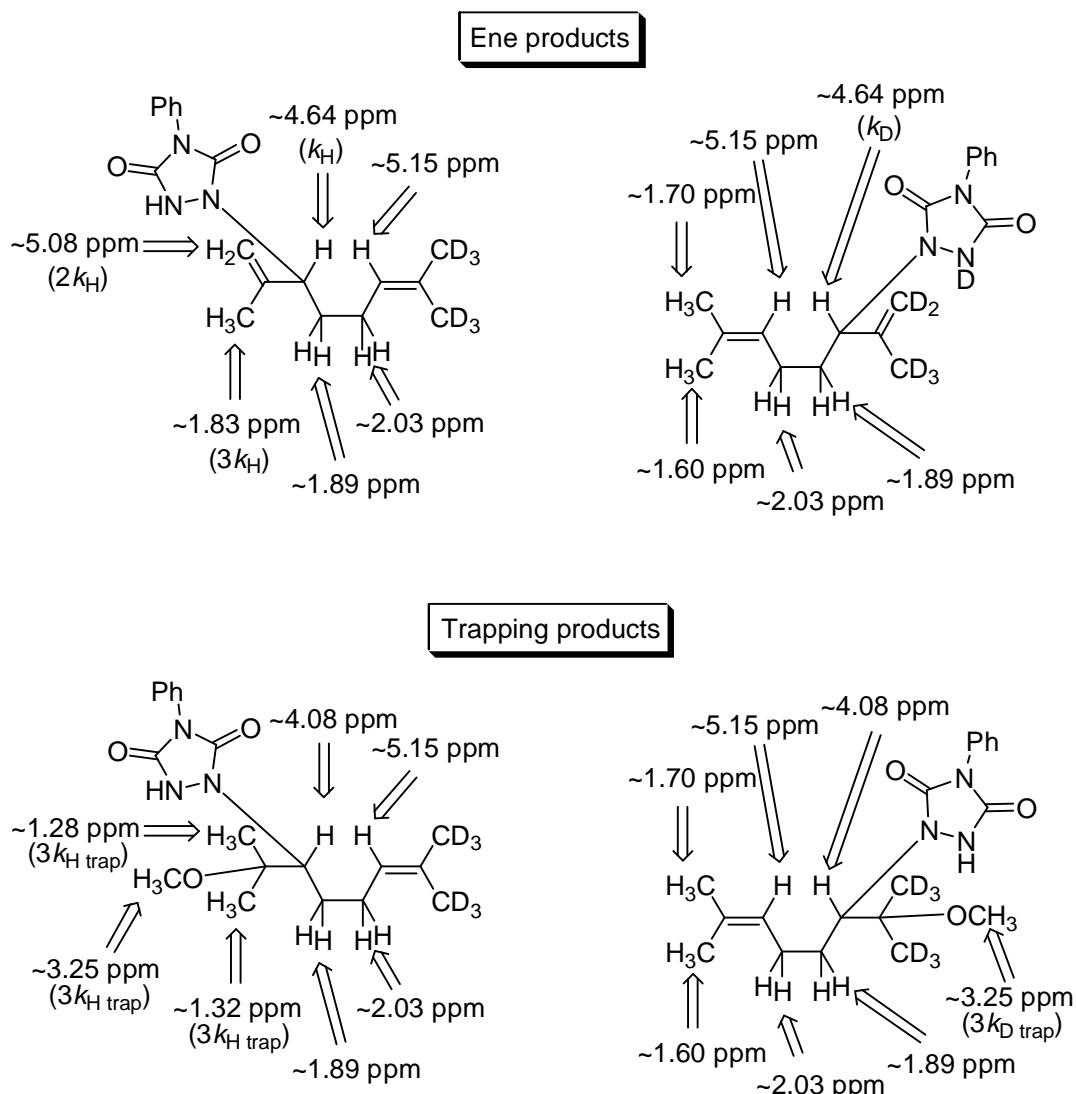


substrate solvent temperature  $k_H/k_D$   
*cis*-TME-*d6* EtOH rt  $2.10 \pm 0.06$





**B)  $^1\text{H}$  NMR spectra for the measurement of the KIEs in the reaction of DMOD- $d_6$  with PTAD**

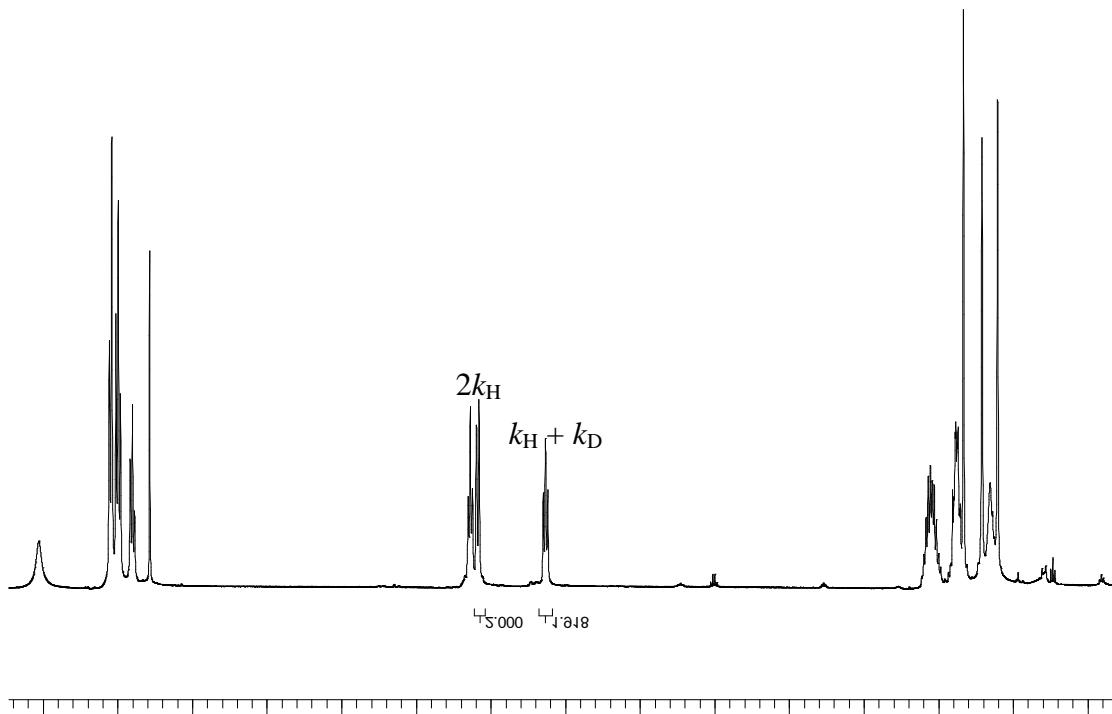


substrate  
DMOD-*d*<sub>6</sub>

solvent  
 $\text{CH}_2\text{Cl}_2$

temperature  
rt

$k_{\text{H}}/k_{\text{D}}$   
 $1.09 \pm 0.03$

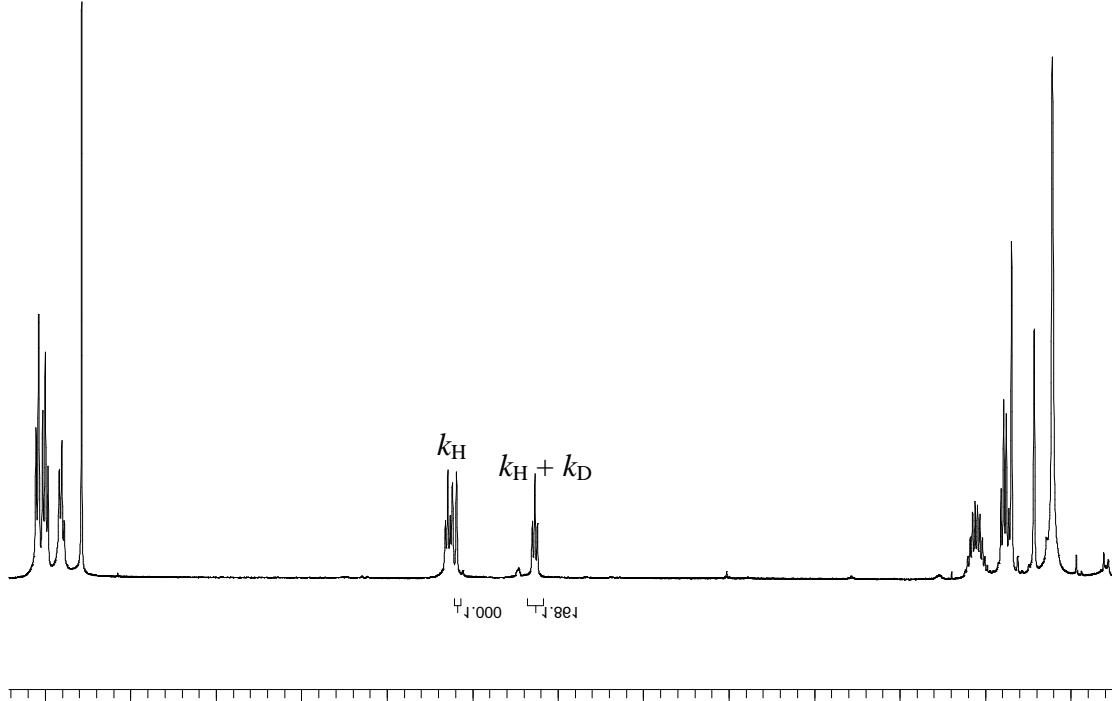


substrate  
DMOD-*d*<sub>6</sub>

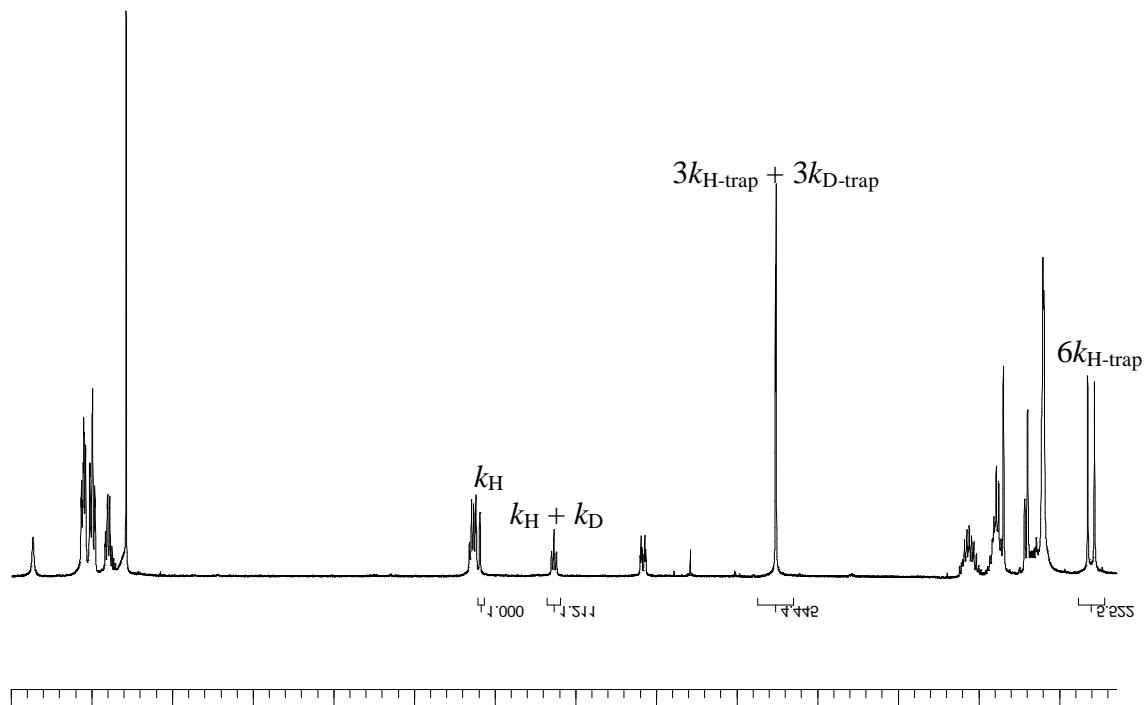
solvent  
 $\text{CH}_2\text{Cl}_2$

temperature  
0 °C

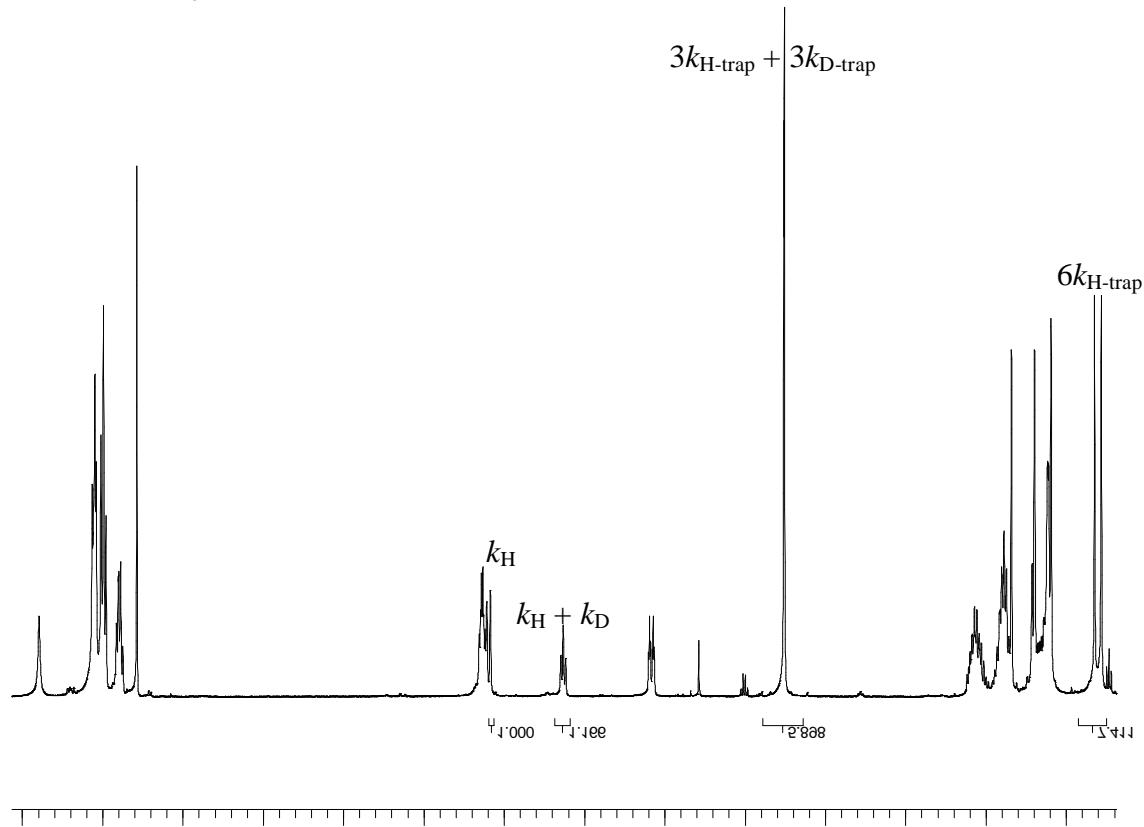
$k_{\text{H}}/k_{\text{D}}$   
 $1.16 \pm 0.03$

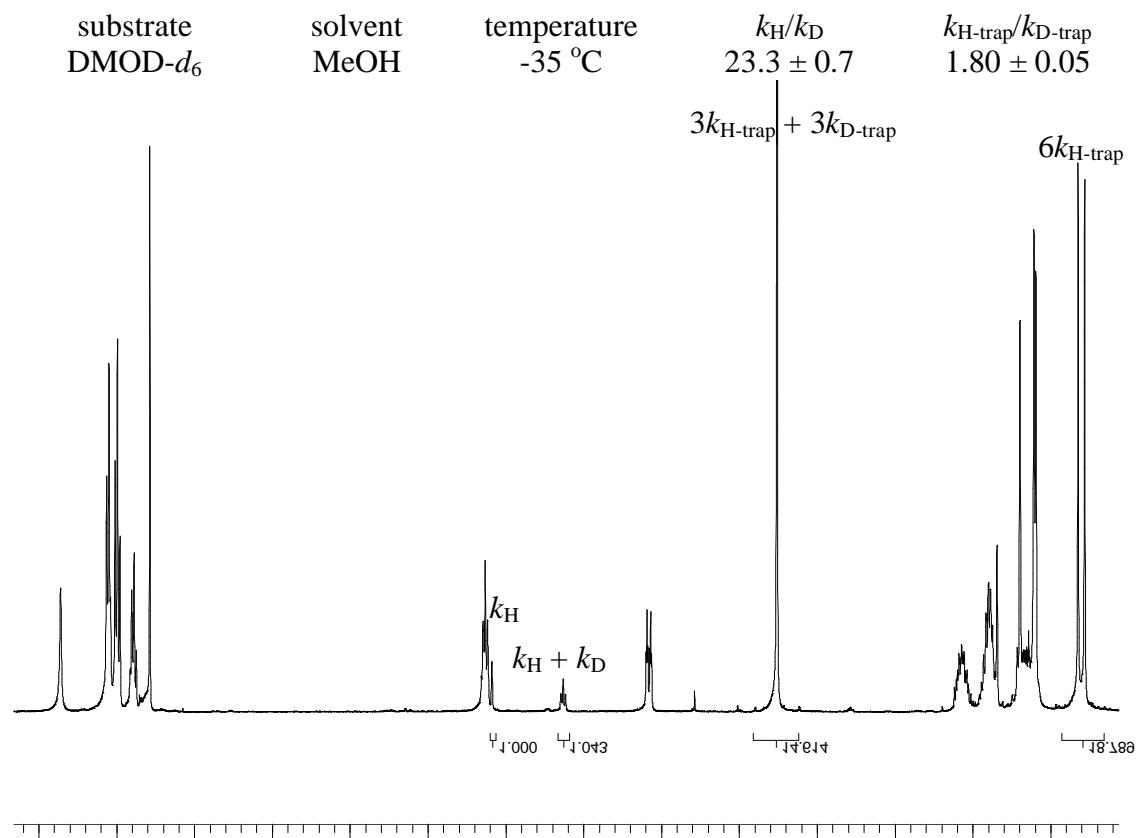
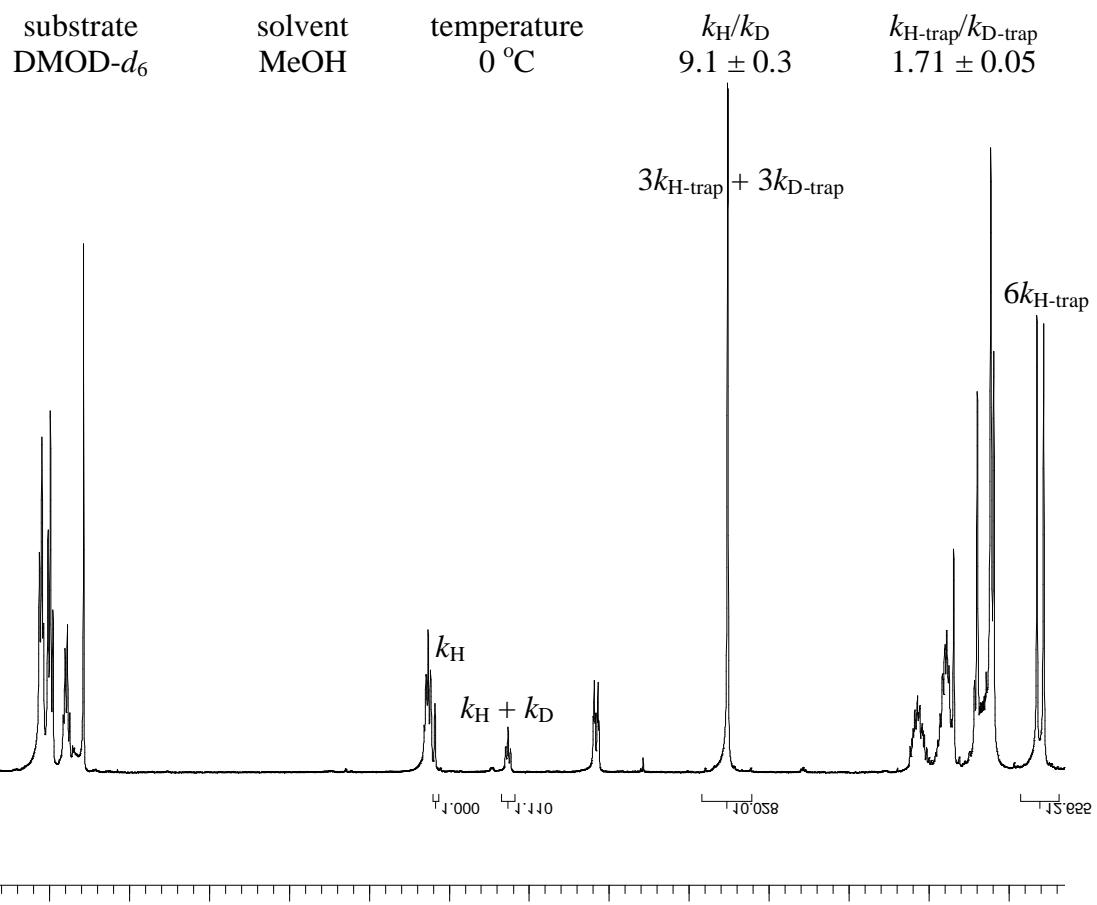


substrate	solvent	temperature	$k_H/k_D$	$k_{H\text{-trap}}/k_{D\text{-trap}}$
DMOD- <i>d</i> <sub>6</sub>	MeOH	40 °C	4.7 ± 0.1	1.64 ± 0.05

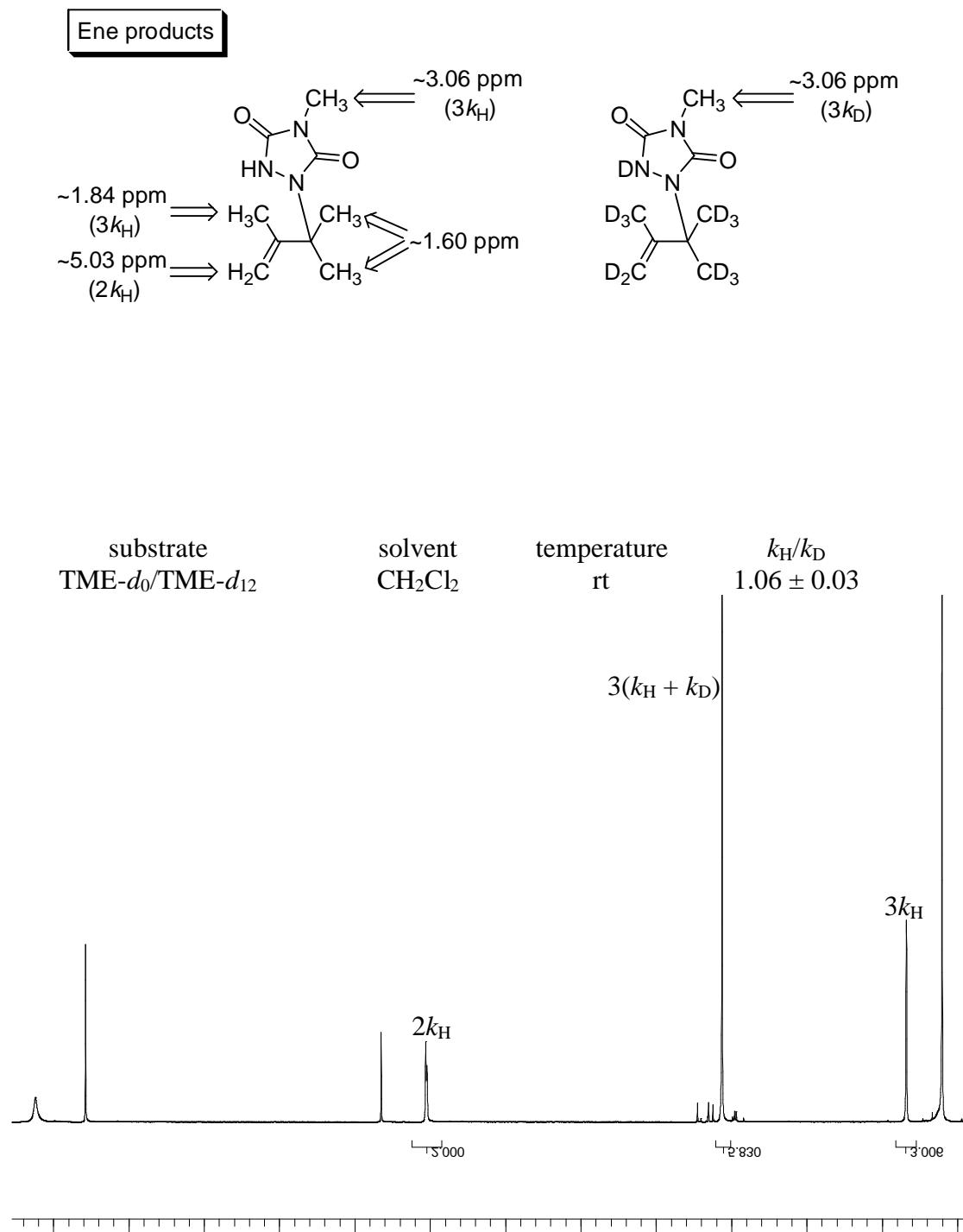


substrate	solvent	temperature	$k_H/k_D$	$k_{H\text{-trap}}/k_{D\text{-trap}}$
DMOD- <i>d</i> <sub>6</sub>	MeOH	rt	6.0 ± 0.2	1.69 ± 0.05

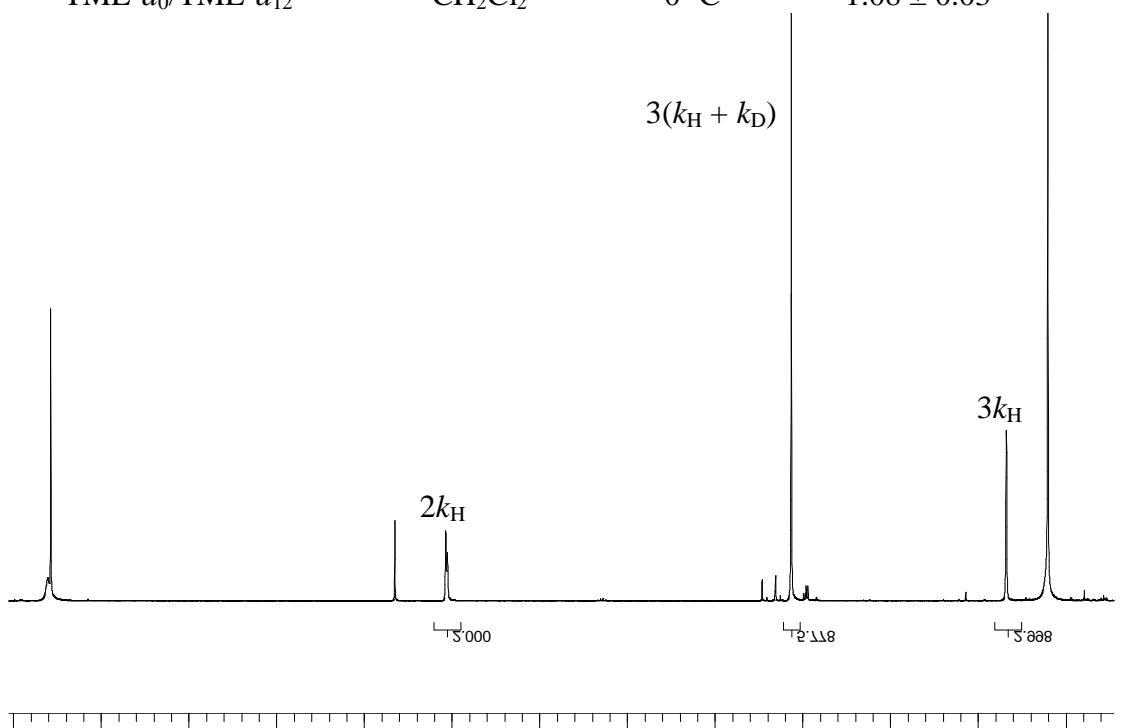




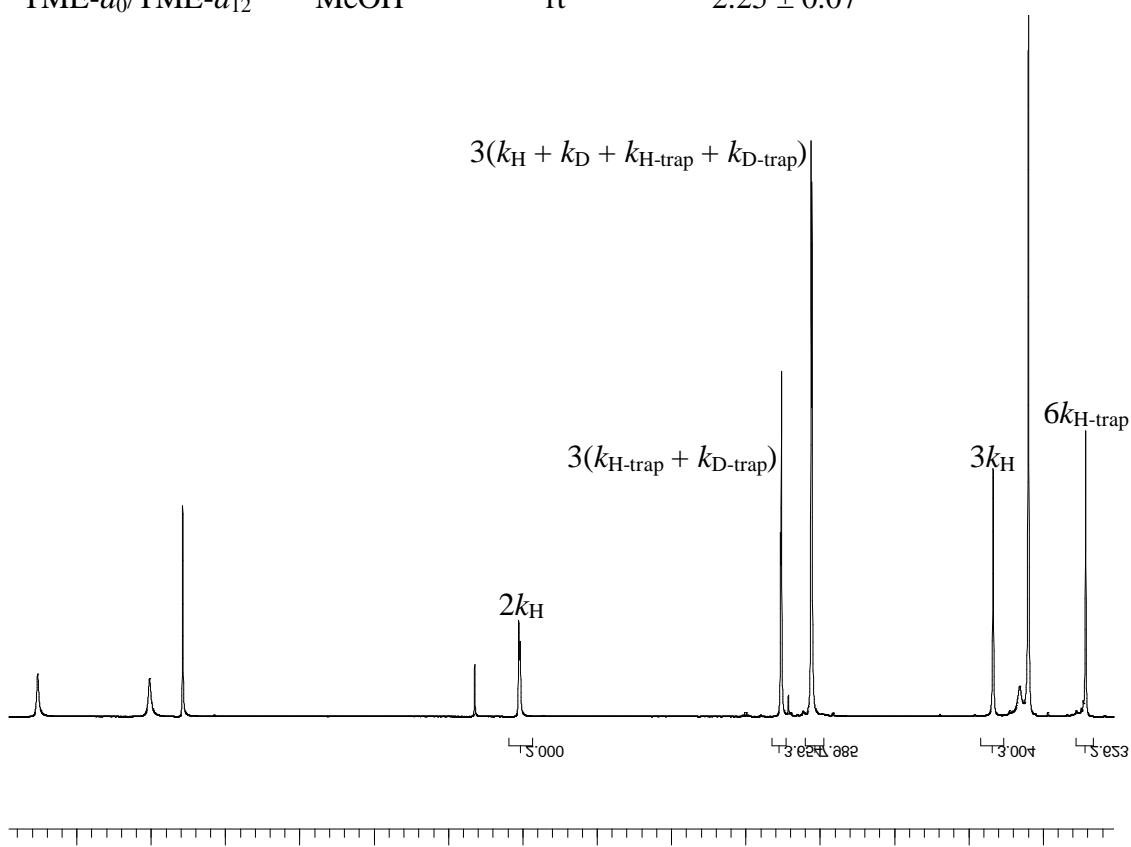
**C)  $^1\text{H}$  NMR spectra for the measurement of the KIEs in the reaction of TME- $d_0$ /TME- $d_{12}$  with MTAD**



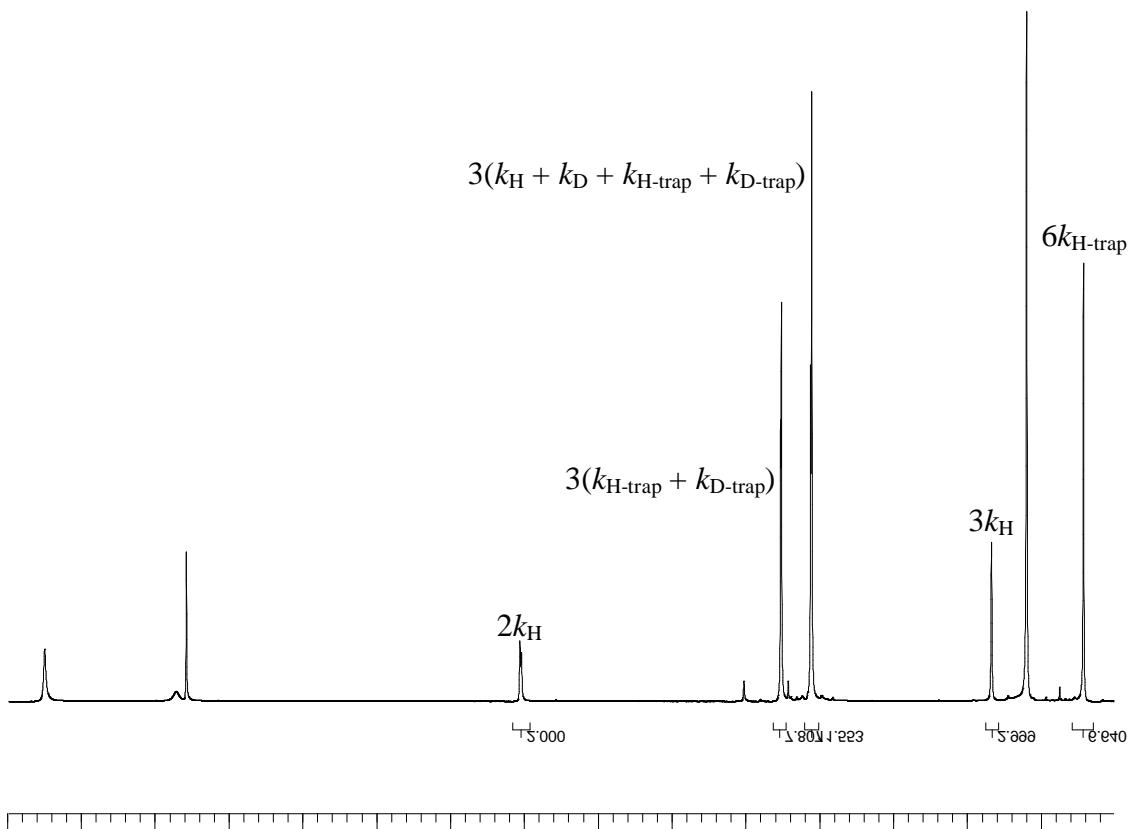
substrate TME-*d*<sub>0</sub>/TME-*d*<sub>12</sub> solvent CH<sub>2</sub>Cl<sub>2</sub> temperature 0 °C  $k_{\text{H}}/k_{\text{D}}$  1.08 ± 0.03



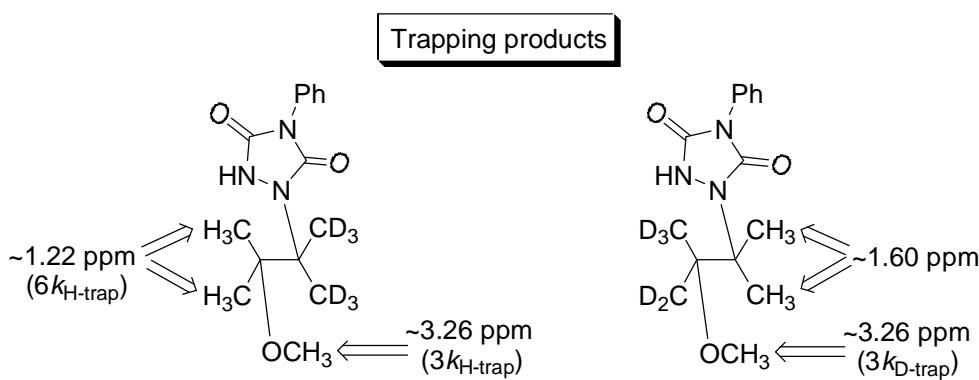
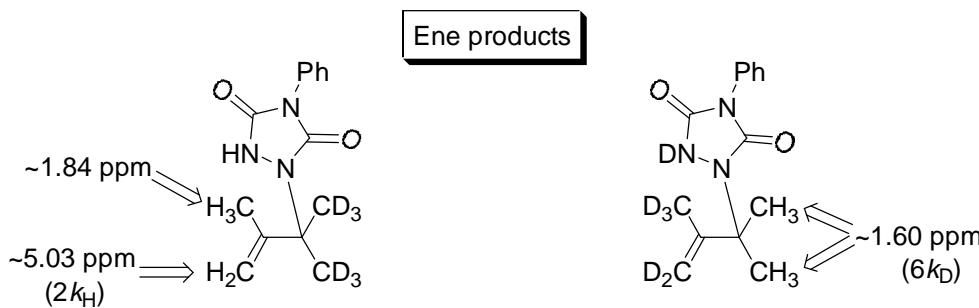
substrate TME-*d*<sub>0</sub>/TME-*d*<sub>12</sub> solvent MeOH temperature rt  $k_{\text{H}}/k_{\text{D}}$  2.25 ± 0.07



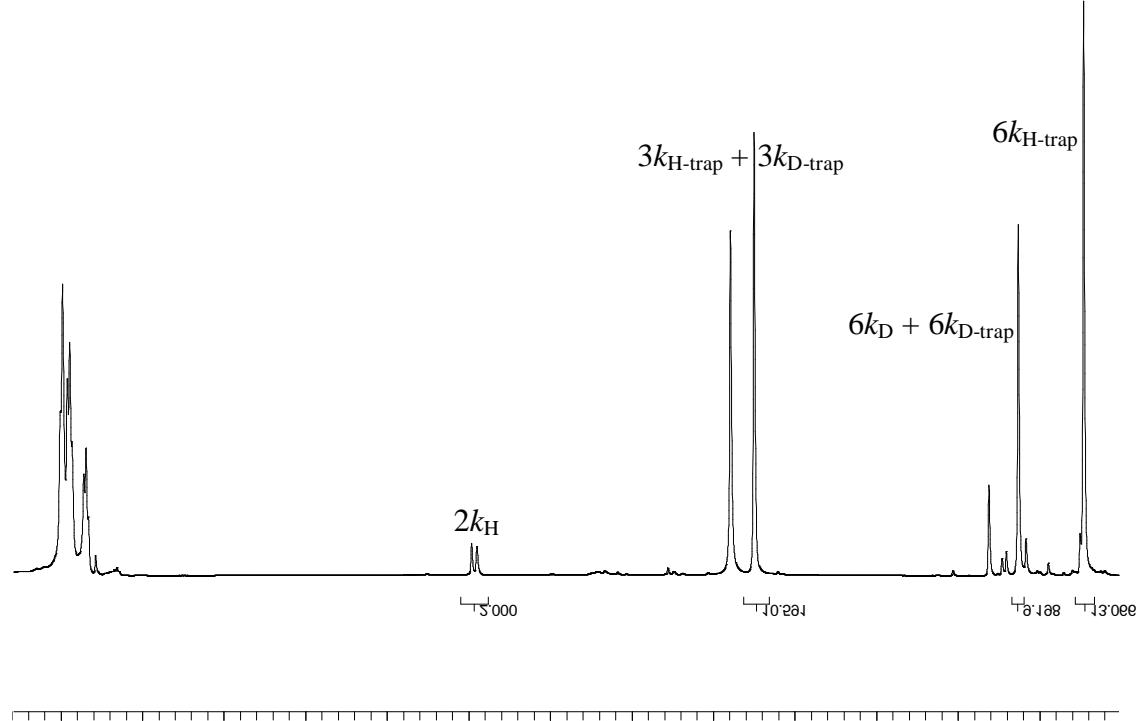
substrate TME-*d*<sub>0</sub>/TME-*d*<sub>12</sub> solvent MeOH temperature 0 °C  $k_{\text{H}}/k_{\text{D}}$   $4.02 \pm 0.12$



**D)  $^1\text{H}$  NMR spectra for the measurement of the KIEs in the reaction of *gem*-TME- $d_6$  with PTAD**



substrate	solvent	temperature	$k_{\text{H}}/k_{\text{D}}$	$k_{\text{H-trap}}/k_{\text{D-trap}}$
<i>gem</i> -TME- $d_6$	MeOH	rt	$5.5 \pm 0.2$	$1.61 \pm 0.05$



substrate	solvent	temperature	$k_H/k_D$	$k_{H\text{-trap}}/k_{D\text{-trap}}$
<i>gem</i> -TME- <i>d</i> 6	MeOH	0 °C	$5.7 \pm 0.2$	$1.64 \pm 0.05$

